

Chapter Test B Arrangement Of Electrons In Atoms

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Modern Chemistry 31 Chapter Test Chapter: Arrangement of Electrons in Atoms PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. ____ 1. The principal quantum number of an electron is 4. What are the

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Modern Chemistry 28 Chapter Test Name Class Date Chapter Test A, continued ____ 8. What is the frequency of a photon whose energy is 3.4 10 19 J? (h 6.626 10 34 J\u00b7s) a. 8.8 1026 Hz b. 5.1 1014 Hz c. 1.9 10-15 Hz d. 2.3 10-52 Hz ____ 9. When electromagnetic radiation strikes the surface of a metal, electrons are ejected from the metal's ...

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(b) particle nature. (d) wave nature. 6. c According to Bohr, which is the point in the figure below where electrons cannot reside? (a) point A (c) point C (b) point B (d) point D 7. c According to the quantum theory, point D in the above figure represents (a) the fixed position of an electron. (b) the farthest position from the nucleus that an ...

4 Arrangement of Electrons in Atoms
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