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provided. 1. Solid CaCl_2 does not conduct electricity. Explain why it is considered to be an electrolyte. CaCl_2 is an ionic solid. In the crystal form, its ions are locked in position. Dissolving

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SECTION 1 SHORT ANSWER 1. c 2. a 3. b 2. a. alcohol
b. water c. the gels 3. The mixture is a colloid. The properties are consistent with those reported in Table 3 on page 404 of the text. The particle size is small, but not too small, and the mixture

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